## **CLAIMS**

- 1. Process for the electrolysis of Al<sub>2</sub>S<sub>3</sub>, using a bath of molten salt, preferably a bath of molten chloride salt, in which Al<sub>2</sub>S<sub>3</sub> is dissolved characterised in that measures are taken to improve the electrical conductivity of the bath, so as to enable an increase in the current density in the bath.
- 2. Process according to claim 1 characterised in that the measures comprise adding an additive to the bath.
- 3. Process according to claim 2 characterised in that the additive comprises, preferably mainly consists of a fluoride compound.
- 4. Process according to claim 3 characterised in that the fluoride compound is cryolite.
  - 5. Process according to claim 4 characterised in that the concentration of the cryolite is in the range of 5-30 wt%, preferably 7-15 wt%, more preferably about 10 wt%.

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- 6. Process according to any of the preceding claims characterised in that the measures comprise enhancing the effective area of an anode extending into the bath by reducing the amount and/or size of gas bubbles covering the anode.
- Process according to any of the preceding claims characterised in that the bath of molten salt mainly comprises alkali metal chlorides, preferably KCl and NaCl.
- 8. Process according to any of the preceding claims characterised in that the bath of molten metal is substantially free of earth alkaline chlorides.

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- 9. Process according to any of the preceding claims characterised in that the electrolysis is carried out at a bath temperature of between 600°C and 850 °C, preferably between 700 °C and 800 °C.
- 5 10. Process according to any of the preceding claims characterised in that the electrolysis is carried out in a multi-polar electrolysis cell.